Practical Synthesis of Oligodehydroalanine Derivatives by Repetition of Stepwise Elongation of Serine Derivative and β -Elimination

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Facile and practical synthesis of various oligodehydroalanine and its amide derivatives, which compose most of the thiostrepton peptide antibiotics, were first accomplished by repetition of the stepwise elongation of serine derivative and then β -elimination.

In recent years, various kinds of thiostrepton peptide antibiotics, which have a characteristic macrocyclic structure, have been isolated from the cultures of various Streptomyces (St.) strains. 1) For example, the structure of similar peptides A10255G and $J_{*}^{(2)}$ obtained from St. gardneri, is illustrated in Fig. l. All of such peptides are characterized by comprising polythiazole and/or polyoxazole-dehydropeptide substructure. Futhermore, interestingly enough, a wide variety of linear oligodehydroalanines or their amide segments link invariably to the macrocyclic peptide ring. So far, the syntheses of the protected dehydroalanine(Δ Ala) by two methods, that is, the β -elimination of serine (Ser) derivatives $^{3,4)}$ and the Hofmann degradation of diaminopropanates, 5) have been reported. However, the effective synthesis of the oligo-dehydroalanine derivative has not been reported, because the formed ΔAla derivative and its peptide are very labile and readily polymerize during the isolation.

BocHN

B

Fig. 2.

i) H-Ser(TBS)-OMe (1), DCC, HOBt, DMF, r.t., 6 h, ii) 70% AcOH, r.t., 12 h, iii) (a) MsCl, Et₃N, CH₂Cl₂, 35 °C, 15 min (b) Sonication, Et₃N, CH₂Cl₂, 35 °C, 15 min, iv) 1M LiOH, H₂O / dioxane, 0 °C, 15 min, v) Repeated twice, three, and four times the coupling of **3b**, **4b**, and **5b** with **1** and β -elimination, vi) (a) DCC, HOSu, THF, 0 °C, 30 min (b) NH₄OH, 0 °C, 10 min.

Scheme 1.

In this paper, we demonstrate a useful synthetic method for various oligodehydroalanines and their amides coupled with the carbonyl groups of picolinic acid (Pic-OH) and (S)-2-(1-t)-butoxycarbonylamino)ethylthiazole-4-carboxylic acid, as shown in Fig. 2.

First, the coupling of Pic-OH with Ser derivatives and subsequent base-catalyzed β -elimination were examined exclusively. That is, H-Ser(TBS)-OMe (1), derived by the O-protection of H-Ser-OMe with t-butyldimethylsilyl chloride (TBSCl), was found to readily couple with Pic-OH in DMF in the presence of 1-hydroxybenzotriazole (HOBt) by the dicyclohexylcarbodiimide (DCC) method to give Pic-Ser(TBS)-OMe (2a) in 80% yield, whereas, in the similar case with H-Ser-OMe, the yield was less than 20%. After removal of TBS group with 70% AcOH, the mesylation of the obtained Pic-Ser-OMe (2b) with methanesulfonyl chloride (MsCl) in the presence of Et₃N yielded Pic-Ser(Ms)-OMe (2c) as the intermediate, which was then subjected to the β -elimination with Et₃N under sonication. As a result, the effective β -elimination of mesyloxy group of

Table 1. The yields and ¹H NMR data of 3-8

Compd No.	Yield / %	¹ H NMR, δ ^{a)} vinyl-H (J _{Hz})	Compd No.	Yield / %	¹ H NMR, δ ^{a)} vinyl-H (J _{Hz})
3a	96	6.02 d (1.8), 6.82 s	6a	71	5.33-5.58 m (3H), 6.03 s, 6.65 s, 6.66 d (2.0), 6.74 d
4a	90	5.48 s, 6.00 s, 6.72 s, 6.78 s			(2.9), 6.82 d (2.0)
5a	82	5.49 dd (0.7, 2.4), 5.56 dd (1.0, 2.0), 6.01 d (0.7), 6.64 s, 6.72 d (2.4), 6.89 d (2.0)	7	40	5.78 s, 6.57 s
			8	45	5.54 s (2H), 6.53 s (2H)

a) Measured in CDCl₃.

2c with Et_3N in CH_2Cl_2 proceeded very smoothly under weak basic conditions to give $Pic-\Delta Ala-OMe$ (3a) in 96% yield without polymerization during the reaction and the purification. Subsequently, hydrolysis of ester 3a with 1M-LiOH gave the corresponding free acid (3b). Thus, similar stepwise elongation of 3b with 1, followed by the consecutive deprotection, mesylation, and β -elimination repeatedly twice, three and four times gave Pic-di (4b; n=2)-, tri (5b; n=3)-, and tetradehydroalanines (6b; n=4) via the corresponding methyl esters (4a, 5a, and 6a), respectively. Furthermore, the amidation of 3b with NH_4OH in the presence of l-hydroxysuccinimide (HOSu) by the DCC method gave the corresponding $Pic-\Delta Ala-NH_2$ (7). The amidation of 4b also gave the expected amide 8 (Scheme 1). The compounds 3-8 purely isolated were found to be very stable. The structures of the all products thus obtained were supported by the 1H NMR spectral data and the satisfactory results of elemental analyses.

In the 1 H NMR spectra of the synthesized oligodehydroalanines (3a, 4a, 5a, and 6a) and their amides (7 and 8), all of the signals of the two, four, six, and eight protons of the vinyl groups in the dehydroalanine residues appeared in the δ =5.48-6.89 region as singlet, doublet, and double doublet (J=0.7-2.4 Hz), as summarized in Table 1.

On the other hand, the synthesis of ethyl (S)-2-(1-BocNH)ethylthiazole-4-carboxylate (11) as another

i) Lawesson's reagent, DME, r.t., 24h, ii) (a) BrCH₂COCOOEt, DME, 0 °C, 15 min (b) TFAA, pyridine, DME, 0 °C, 30 min, iii) 1M LiOH, H₂O / dioxane, r.t., 6 h, iv) 1, DCC, HOBt, DMF, r.t., 6 h, v) 70 % AcOH, r.t., 12 h, vi) (a) MsCl, Et₃N, CH₂Cl₂, 35 °C, 15 min (b) Sonication, Et₃N, CH₂Cl₂, 35 °C, 15 min, vii) Repeated the coupling of 13b with 1 and then β -elimination, viii) (a) DCC, HOSu, THF, 0 °C, 30 min (b) NH₄OH, 0 °C, 10 min.

substrate in the synthesis of its thiazole-4-carbonyl-oligodehydroalanines was examined, according to the method of Schmidt et al. 6) Namely, the thiocarbonylation of Boc-L-Ala-NH2 with Lawesson's reagent gave the corresponding thioamide. Subsequent cyclization of the thioamide with ethyl bromopyruvate in dimethylethane (DME) in the presence of KHCO3, followed by the oxidation with trifluoroacetic acid anhydride (TFAA) in the presence of pyridine gave (S)-11a. (7,8)

After ester hydrolysis with 1 M-LiOH, as in the case of 2a-c, the obtained free acid [11b: Mp 128-129 °C. δ =8.32 (s, 1H, ring-H), 12.87 (br s, 1H, -COOH)] was coupled with 1 to give dipeptide (12a), which was successively deprotected [12b: Mp 142-143 °C. δ =2.90 (br s, 1H, -OH), 8.04 (s, 1H, ring-H)], mesylated (12c), and then the mesyloxy group was eliminated to give 2-(1-BocNH)ethylthiazole-4-carbonyl-ΔAla-OMe [13a: Mp 120-121 °C. δ =5.79 (d, 1H, vinyl-H, J=1.5 Hz), 6.57 (s, 1H, vinyl-H), 8.05 (s, 1H, ring-H)].

Furthermore, after ester hydrolysis with 1 M-LiOH, stepwise elongation of the obtained acid [13b, Mp 148-149 °C. δ =5.80 (s, 1H, vinyl-H), 6.50 (s, 1H, vinyl-H), 10.50 (br s, 1H, -COOH)] with **1** and then β elimination were repeated to give 2-(1-BocNH)ethylthiazole-4-carbonyl-didehydroalanines 14a [Mp 121-122 °C. δ = 5.41 (t, 1H, vinyl-H, J=2.0 Hz), 5.94 (d, 1H, vinyl-H, J=1.1 Hz), 6.62 (s, 1H, vinyl-H), 6.67 (d, 1H, vinyl-H, J=2.2 Hz) and 14b [Mp 183-184 °C. $\delta=5.79$ (s, 1H, vinyl-H), 5.84 (s, 1H, vinyl-H), 6.00 (s, 1H, vinyl-H), 6.45 (s, 1H, vinyl-H), 8.32 (s, 1H, ring-H), 12.50 (br s, 1H, -COOH)] (Scheme 2). Finally, the amidations of both 13b and 14b with NH, OH by the DCC method were similarly carried out to give the expected amides 15 $(n=1)^9$ and 16 $(n=2)^9$ in 44% and 38% yields, respectively.

In conclusion, it is noteworthy that the synthesis of stable oligodehydroalanine derivatives were first achieved, and this synthetic method is applicable to the synthesis of the other dehydropeptides containing a dehydroamino acid residue.

References

- 1) Y. Egawa, K. Umino, Y. Tamura, M. Shimizu, K. Kaneko, M. Sakurazawa, S. Awataguchi, and T. Okuda, J. Antibiot., 22, 12 (1969); N. Nakanishi, T. Oshida, S. Yano, K. Takeda, T. Yamaguchi, and Y. Ito, Plasmid, 15, 217 (1986); M. Matsumoto, T. Kawamura, Y. Yasuda, T. Tanimoto, K. Matsumoto, T. Yoshida, and J. Shoji, J. Antibiot., 42, 1465 (1989).
- 2) M. Debono, R. M. Molloy, J. L. Occolowitz, J. W. Paschal, A. H. Hunt, K. H. Michel, and J. W. Martin, J. Org. Chem., 57, 5200 (1992); M. E. Favret and L. D. Boeck, J. Antibiot., 45, 1809 (1992).
- 3) I. Photaki, J. Am. Chem. Soc., 85, 1123 (1963).
- 4) A. Srinivasan, R. W. Stephenson, and R. K. Olsen, J. Org. Chem., 42, 2253 (1977).
- 5) S. Nomoto, A. Sano, and T. Shiba, Tetrahedron Lett., 1979, 521.

- 6) U. Schmidt, P. Gleich, H. Griesser, and R. Utz, *Synthesis*, **1986**, 992. 7) (*R*)-**11a**: Mp 89.5 °C. $[\alpha]_D^{20}$ 38.6° (c 0.67, CH₂Cl₂). 6 8) (*S*)-**11a**: Mp 87-88 °C. $[\alpha]_D^{26}$ -43.2° (c 0.67, CH₂Cl₂). δ =1.40 (t, 3H, -CH₂CH₃, *J*=7.0Hz), 1.44 (s, 9H, Boc), 1.62 (d, 3H, -CH₃, *J*=6.8 Hz), 4.42 (q, 2H, -CH₂CH₃, *J*=7.0 Hz), 4.90-5.35 (m, 2H, NH and
- α-H), 8.08 (s, 1H, ring-H). 9) **15**: Mp 138-139 °C. [α]_D²⁶ -30.0° (c 0.14, MeOH). δ=1.45 (s, 9H, Boc), 1.53 (d, 3H, CH₃, J=7.0 Hz), 4.85-5.01 (m, 1H, α-H), 5.76 (s, 1H, vinyl-H), 6.51 (s, 1H, vinyl-H), 7.67 (br s, 1H, NH₂'s NH), 7.89 (br d, 1H, Ala's NH J=7.9 Hz), 8.14 (br s, 1H, NH₂'s NH), 8.34 (s, 1H, ring-H), 9.94 (br s, 1H, Δ Ala's NH). **16:** Mp 159-160 °C. $[\alpha]_D^{26}$ -3.45° (c 0.29, MeOH). δ =1.43 (s, 9H, Boc), 1.53 (d, 3H, CH₃, J=7.0 Hz), 4.85-5.01 (m, 1H, α -H), 5.73 (s, 2H, vinyl-H), 6.05 (s, 1H, vinyl-H), 6.44 (s, 1H, vinyl-H), 7.52 (br s, 1H, NH₂'s NH), 7.94 (br s, 1H, NH₂'s NH), 8.10-7.75 (m, 1H, Ala's NH), 8.35 (s, 1H, ring-H), 9.45 (br s, 1H, Δ Ala's NH), 9.89 (br s, 1H, Δ Ala's NH).

(Received March 28, 1994)